

GANODERIC ACID DERIVATIVES AND ERGOSTA-4,7,22-TRIENE-3,6-DIONE FROM *GANODERMA LUCIDUM**

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Key Word Index—*Ganoderma lucidum*; Basidiomycetes; cultured mycelia; triterpenoids; ganoderic acids T; S; R; Q; P; O; ergosta-4,7,22-triene-3,6-dione.

Abstract—Seven new triterpenoids, ganoderic acid T, ganoderic acid S, ganoderic acid R, ganoderic acid P, ganoderic acid Q, ganoderic acid O, 7-O-methyl-ganoderic acid O and a known ergosterol derivative, ergosta-4,7,22-triene-3,6-dione were isolated from the cultured mycelium of *Ganoderma lucidum*. The structure of the first compound was determined using spectroscopic and X-ray analysis, and the structures of the other compounds were elucidated by spectroscopic data.

INTRODUCTION

Ganoderic acid derivatives, highly oxygenated and pharmaceutical active lanostane type triterpenoids, were recently isolated from the fungus *Ganoderma lucidum* (Polyporaceae) [1–15]. Toth *et al.* [16, 17] isolated ganoderic acids Z, Y, X, W, V and U, which were cytotoxic to hepatoma cells *in vitro*, from the cultured mycelium of *G. lucidum*. In a preliminary paper [18], we have reported the isolation and structure elucidation of three new ganoderic acid derivatives, ganoderic acid T(2a), S(2b) and R(2c) as the major triterpene constituents of the cultured mycelium. In this paper we wish to deal with the isolation and structure elucidation of these ganoderic acid derivatives as major triterpenoids and further new ganoderic acid derivatives and a new fungal metabolite, ergosta-4,7,22-triene-3,6-dione as a minor component of the cultured mycelium of *G. lucidum*.

RESULTS AND DISCUSSION

The chloroform extract (16.78 g) of the cultured mycelia was chromatographed on silica gel and separated into fractions A–E. Further purification by reversed-phase HPLC afforded compounds 1, 2a, 2b, 2c, 2d, 2e and 3b from fractions B, C and E.

Compound 1, ergosta-4,7,22-triene-3,6-dione, analysed for $C_{28}H_{40}O_2$ and showed in the UV spectrum an absorption at 275 ($\log \epsilon$ 3.8) characteristic of an $\alpha,\beta,\alpha',\beta'$ -unsaturated carbonyl group. The IR spectrum of 1 showed the presence of an α,β -unsaturated carbonyl group ($1665, 1620\text{ cm}^{-1}$) and an $\alpha,\beta,\alpha',\beta'$ -unsaturated carbonyl group ($1640, 1600\text{ cm}^{-1}$). The ^1H NMR spectrum of 1 (Table 1) clearly showed the presence of two tertiary methyl and four secondary methyl groups, thus

suggesting an ergostane skeleton. This was supported by the base peak ion in the mass spectrum at m/z 283 (100%) corresponding to the fragment obtained by the loss of the side chain resulting from cleavage between C-17 and C-20. Furthermore, in the ^1H NMR spectrum of 1, the chemical shift values of this protons of four secondary methyl groups were completely consistent with those of ergosterol [19]. Thus, compound 1 was suggested to have the same side chain as ergosterol. This suggestion was supported by the ^1H NMR spectrum, which showed two coupled olefinic proton signals (δ 5.14 and 5.24). On the other hand, two olefinic protons (δ 5.98 *br s* and 6.47 *s*) were observed in the ^1H NMR spectrum of 1. Also four tertiary carbon signals (δ 123.9, 126.0, 132.8, 134.8), two quaternary carbon signals (δ 158.2 and 167.7) and two carbonyl carbon signals (δ 187.0 and 199.3) were observed in the ^{13}C NMR spectrum of 1. All these data indicated that 1 was an ergostane derivative with two trisubstituted olefins, an α,β -unsaturated carbonyl group and an $\alpha,\beta,\alpha',\beta'$ -unsaturated carbonyl group in the ergostane skeleton. By comparison of the ^1H NMR spectral data of compound 1, ergosterol and $3\beta,5\alpha,14\alpha$ -trihydroxyergosta-7,22-dien-6-one, the C-18 methyl proton signals were observed at δ 0.68, 0.63 and 0.68, respectively, but the chemical shift values of the C-19 methyl proton signals were very different (Table 1). These data led to the conclusion that the functional groups described above were present in the A and B ring portion of the molecule. Furthermore, the olefinic proton (δ 6.47) was not coupled with any other proton. Therefore, the structure of 1 was elucidated as ergosta-4,7,22-triene-3,6-dione. Malorini *et al.* [21] reported the occurrence of ergosta-4,7,22-triene-3,6-dione in a mixture of steroidal $\Delta^{4,7}$ -3,6-diketones from the extract of the marine sponge *Raphidostila incisa*. The isolation of ergosta-4,7,22-triene-3,6-dione from the cultured fungi has been demonstrated for the first time [22].

Compound 2a, ganoderic acid T, analysed for $C_{36}H_{52}O_8$ and showed mp 200–202°, and absorptions in

*Part 7 in the series 'Studies on the Metabolites of Higher Fungi'. For Part 6 see ref. [18].

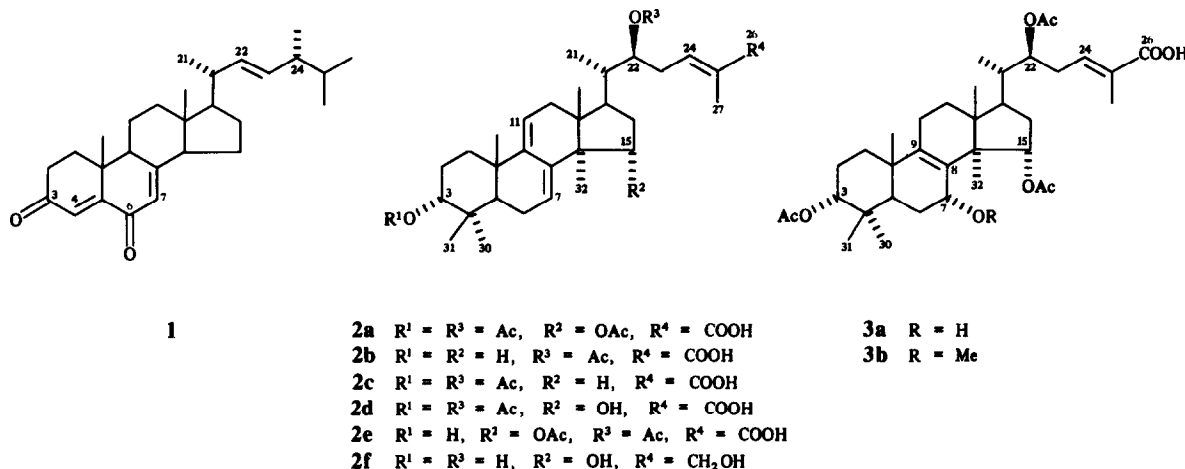


Table 1. ^1H NMR spectral data of compound 1, ergosterol and 3 β ,5 α ,14 α -trihydroxyergosta-7,22-dien-6-one

Proton	Compound 1	Ergosterol	3 β ,5 α ,14 α -Trihydroxy-ergosta-7,22-dien-6-one
18-H ₃	0.68	0.63	0.68
19-H ₃	1.30	0.95	0.94
21-H ₃	1.04 (<i>d</i> , <i>J</i> = 6.6 Hz)	1.04 (<i>d</i> , <i>J</i> = 6 Hz)	1.02 (<i>d</i> , <i>J</i> = 6.2 Hz)
26 and 27-H ₃	0.82 (<i>d</i> , <i>J</i> = 6.6 Hz)	0.82 (<i>d</i> , <i>J</i> = 6 Hz)	0.82 (<i>d</i> , <i>J</i> = 6.6 Hz)
	0.84 (<i>d</i> , <i>J</i> = 6.6 Hz)	0.84 (<i>d</i> , <i>J</i> = 6 Hz)	0.84 (<i>d</i> , <i>J</i> = 6.6 Hz)
28-H ₃	0.92 (<i>d</i> , <i>J</i> = 6.6 Hz)	0.92 (<i>d</i> , <i>J</i> = 6 Hz)	0.92 (<i>d</i> , <i>J</i> = 6.6 Hz)

The data for ergosterol and 3 β ,5 α ,14 α -trihydroxyergosta-7,22-dien-6-one are cited from refs [19, 20].

the UV spectrum at 234, 242 and 250 nm, suggesting the presence of a heteroannular diene moiety. The IR spectrum of 2a showed the presence of a hydroxyl group (3420 cm^{-1}) and ester carbonyl group (1720 cm^{-1}). The ^1H NMR spectrum of 2a showed the presence of five tertiary methyl (δ 0.66, 0.88, 0.98, 0.99 and 1.03), one allyl methyl (δ 1.86) and one secondary methyl (δ 0.97, *d*, *J* = 6.7 Hz) signals. Furthermore, the signals due to three secondary acetoxy groups and three olefinic proton signals were observed (Table 2). These spectral data suggested 2a to be a lanostane type triterpenoid having a heteroannular diene moiety and three acetoxy groups. The ^{13}C NMR spectral data of 2a supported this suggestion and revealed the presence of one carboxyl group (Table 3). Comparison of the ^1H NMR and ^{13}C NMR spectral data of 2a with those of methyl ganoderate X and 3-*O*-acetyl methyl ganoderate X[17] led to the conclusion that 2a is a tri-acetoxy-ganoderic acid of the 7,9(11)-24-triene type. Two of the three acetoxy were proved to be at C-3 α and C-15 α on the basis of the chemical shift values and the coupling patterns of the methine proton signals which corresponded to those due to the 3 β -H and 15 β -H of 3-*O*-acetyl ganoderate X[17]. The third acetyl group was located at C-22 by reference to the ^1H - ^1H shift correlation NMR spectrum, in which correlation between 23-H [δ 2.33(1H, *ddd*, *J* = 15.0, 7.5, 7.0 Hz) and 2.57(1H, *ddd*, *J* = 15.0, 7.6, 7.1 Hz)] and the third acetoxy methine proton signal at δ 5.03 was observed. The relative stereo-

structure of 2a was determined by X-ray analysis. Crystal data: $M_r = 612.8$, orthorhombic, space group $P2_12_12_1$, $a = 16.740(1)$, $b = 29.115(3)$, $c = 7.213(1)$ Å, $V = 3515.6(6)$ Å³, $z = 4$, $d(\text{calcd}) = 1.157 \text{ g cm}^{-3}$, $d(\text{obsd}) = 1.13 \text{ g cm}^{-3}$ (floatation). The structure was solved by MULTAN 84 and refined by a block-diagonal least-squares technique to $R = 0.057$ for 2130 reflections [$|F| > (Fo)$ and $|\Delta F| < 3\sigma(Fo)$] out of 2994 unique ones obtained in the range of $\theta \leq 60^\circ$ for CuK α radiation. The structure represented Fig. 1 was drawn using the PLUTO program [23].

The absolute stereostructure was determined by the CD spectrum of compound 2f prepared from 2a by reduction, which showed a positive Cotton effect ($[\theta]_{239} + 12689$, $c = 0.012$ EtOH) like that of (26*S*)-26-*O*-methyl-perenniporiol ($[\theta]_{240} + 29824$, $c = 0.012$ EtOH) [24]. Thus, the structure of 2a was determined as (22*S*, 24*E*)-3 α ,15 α ,22-triacetoxy-5 α -lanosta-7,9(11),24-trien-26-oic acid.

Compound 2c, ganoderic acid R, $\text{C}_{34}\text{H}_{50}\text{O}_6$, showed mp 201–202° and it had UV absorptions at 234, 242, 250 nm and hydroxyl (3400 cm^{-1}), ester carbonyl (1720 cm^{-1}) and α,β -unsaturated carbonyl (1675 cm^{-1}) absorptions in the IR spectrum. The ^1H NMR spectrum of 2c was very similar to that of 2a except for the loss of the methine proton signal at δ 5.08 (15 α -H) and of one acetoxy methyl proton signal. The ^{13}C NMR spectrum of 2c was also closely similar to that of 2a except for the appearance of a methylene carbon signal (δ 31.3) instead

Table 2. ^1H NMR spectral data of compounds 2a-2e, 3a and 3b (400 MHz, CDCl_3)*

Proton	2a	2b	2c	2d	2e	3b	3a†
18- H_3	0.66 s	0.57 s	0.57 s	0.60 s	0.65 s	0.70 s	0.68 s
19- H_3	0.99 s	1.00 s	1.05 s	0.99 s	0.98 s	0.94 s	0.92 s
21- H_3	0.97 d (6.7)	0.98 d (6.7)	0.98 d (6.7)	0.96 d (6.7)	0.96 d (6.7)	0.96 d (6.7)	1.00 d (6.7)
30- H_3	0.88 s	0.98 s	0.87 s	0.87 s	0.98 s	0.89 s	1.17 s
31- H_3	0.98 s	0.94 s	0.99 s	0.98 s	0.93 s	0.98 s	0.89 s
32- H_3	1.03 s	0.87 s	0.90 s	1.00 s	0.99 s	1.18 s	1.32 s
27- H_3	1.86 d (1.6)	1.86 d (1.6)	1.87 d (1.6)	1.86 d (1.2)	1.86 d (1.4)	1.86 s	1.94 s
22-OAc	2.05 s	2.05 s	2.05 s	2.05 s	2.06 s	2.05 s	1.76 s
3-OAc	2.07 s	—	2.06 s	2.07 s	2.09 s	2.07 s	1.77 s
15-OAc	2.08 s	—	—	—	—	2.09 s	1.85 s
H-3 β	4.68 dd (3.0, 3.0)	3.45 dd (3.0, 3.0)	4.67 dd (3.0, 3.0)	4.67 dd (2.5, 2.5)	3.45 dd (2.7, 2.7)	4.68 dd (2.3, 2.3)	5.03 dd (2.5, 2.5)
7-H	5.48 brs	5.47 brs	5.47 br dd (4.6, 4.6)	5.88 brd (5.7)	5.48 dd (3.3, 3.3)	3.56 brs	4.30 brd (3.1)
11-H	5.32 d (6.4)	5.33 d (6.4)	5.32 d (6.4)	5.31 brd (5.9)	5.33 br d (6.2)	—	—
15-H	5.08 dd (8.0, 4.0)	—	—	4.27 dd (9.6, 4.9)	5.07 dd (8.8, 5.0)	5.11 dd (7.6, 3.5)	5.53 dd (9.5, 6.0)
22-H	5.03 ddd (7.1, 7.0, 1.4)	5.10 ddd (6.8, 6.0, 1.7)	5.11 ddd (7.1, 7.0, 1.4)	5.04 dd (7.2, 7.2)	5.02 dd (7.6, 7.6)	5.02 dd (7.5, 7.5)	5.30 dd (8.0, 5.5)
23- H_a	2.33 ddd (15.0, 7.5, 7.0)	2.36 ddd (15.0, 7.5, 6.0)	2.37 ddd (15.0, 7.5, 7.0)	2.36 ddd (15.0, 7.3, 7.2)	2.33 ddd (15.0, 7.6, 7.3)	2.33 ddd (15.2, 7.5, 7.5)	2.08 ddd (15.0, 7.5, 5.5)
23- H_b	2.57 ddd (15.0, 7.6, 7.1)	2.56 ddd (15.0, 7.6, 7.0)	2.57 ddd (15.0, 7.6, 7.1)	2.56 ddd (15.0, 7.3, 7.2)	2.57 ddd (15.0, 7.6, 7.3)	2.57 ddd (15.2, 7.5, 7.5)	2.46 ddd (15.0, 8.0, 7.5)
24-H	6.78 ddq (7.6, 7.5, 1.6)	6.82 ddq (7.6, 7.5, 1.6)	6.84 ddq (7.6, 7.5, 1.6)	6.80 ddq (7.3, 7.3, 1.2)	6.78 ddq (7.3, 7.3, 1.4)	6.78 dd (7.5, 7.5)	7.11 ddq (7.5, 7.5, 1.3)
O-Me	—	—	—	—	—	3.16 s	—

*Values in parentheses are coupling constants in Hz.

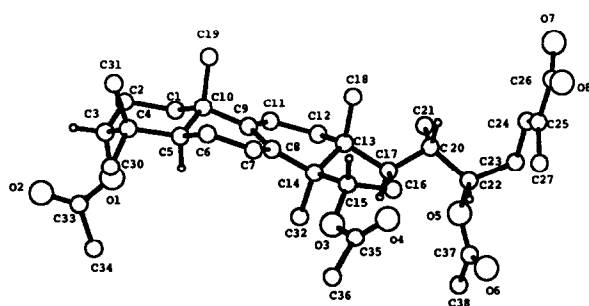
† ^1H NMR spectrum of 3a was taken in C_6D_6 .

Fig. 1. A computer-generated perspective drawing of 2a.

of the methine carbon signal ($\delta 77.4$) seen in the spectrum of 2a. These spectral data showed the structure of 2c to be (22*S*, 24*E*)-3 α ,22-diacetoxy-5 α -lanosta-7,9(11),24-trien-26-oic acid.

Compound 2b, ganoderic acid S, $\text{C}_{32}\text{H}_{48}\text{O}_5$, showed mp 194–196° and it had UV absorptions at 231, 240, 248 nm and hydroxyl (3410 cm^{-1}), ester carbonyl

(1720 cm^{-1}) and α,β -unsaturated carbonyl (1675 cm^{-1}) absorptions in the IR spectrum. The ^{13}C NMR spectrum of 2b was very similar to that of 2c, except for the upfield shift of the C-3 carbon signal ($\delta 76.0$) relative to that of 2c ($\delta 78.0$) and the loss of one acetoxy group. In the ^1H NMR spectrum of 2b, a methine proton signal at $\delta 5.10$ (ddd, $J = 6.8, 6.0, 1.7\text{ Hz}$) due to 22-H was observed and a proton signal at $\delta 3.45$ (dd, $J = 3.0, 3.0, 3\beta\text{-H}$) which was absent in those of 2a and 2c appeared. Based on the spectral evidence, 2b was deduced to be (22*S*, 24*E*)-22-acetoxy-3 α -hydroxy-5 α -lanosta-7,9(11),24-trien-26-oic acid. Fujimoto *et al.* [13] reported the isolation of a new ganoderic acid derivative from the same fungus which was designated ganoderic acid S. After discussion with Dr. Fujimoto, their new compound has been designated as ganoderic acid S₁.

Compound 2d, ganoderic acid Q, mp 131–132°, analysed for $\text{C}_{34}\text{H}_{50}\text{O}_7$ and showed absorptions in the UV spectrum at 235, 243, 251 nm (log ϵ 4.3, 4.3, 4.1), suggesting the presence of a heteroannular diene moiety similar to that in compounds 2a, 2b and 2c. The IR spectrum of 2d showed the presence of absorptions for hydroxyl

Table 3. ^{13}C NMR spectral data of compounds **2a**–**2e**, **3a** and **3b** (100.6 MHz, CDCl_3)*

Carbon	2a	2b	2c	2d	2e	3b	3a†
1	30.8	29.8	30.5	30.6	29.9	30.0	31.3
2	23.3	25.5	23.0	23.1	25.5	23.3	24.3
3	78.2	76.0	78.0	78.1	76.1	77.2	77.7
4	36.8	37.1 ^a	36.4	36.5	37.0	36.3	37.2
5	44.1	43.1	44.0	45.0	42.9	39.6	40.8
6	22.7	27.5	27.5	22.8	23.0	21.3	28.5
7	121.6	120.3	120.2	121.6	121.5	75.9	67.1
8	140.2	142.2	142.2	140.5	140.0	142.9	141.7
9	146.2	145.9	145.8	146.2	146.0	132.7	135.1
10	37.5	37.3 ^a	37.1	37.3	37.3	38.7	39.2
11	115.5	115.5	115.5	115.3	115.3	21.1	21.5
12	38.1	37.6	37.6	38.4	37.9	30.7	32.1
13	44.1	43.6	43.6	44.2	43.9	45.6	46.0
14	51.6	50.3	50.3	52.0	51.4	51.7	52.4
15	77.4	31.2	31.3	74.5	76.9	75.2	76.0
16	36.7	22.9	22.8	39.8	36.7	37.1	37.3
17	45.6	47.3	47.3	45.0	45.4	45.2	46.9
18	15.9	15.4	15.4	15.8	15.8	16.1	17.0
19	23.0 ^a	22.7 ^b	22.4 ^a	22.6 ^a	22.7 ^a	17.6	18.0
20	39.8	39.3	39.3	39.3	39.6	39.9	41.0
21	12.8	12.6	12.6	12.8	12.7	12.7	13.6
22	74.6	74.6	74.6	74.6	74.4	74.3	74.8
23	32.1	31.7	31.8	31.7	31.9	31.8	32.8
24	139.2	139.5	139.6	139.2	139.1	139.1	140.0
25	129.4	129.0	129.0	129.4	129.2	129.3	130.2
26	172.0	172.2	172.3	172.0	171.6	172.2	172.9
27	12.4	12.2	12.2	12.3	12.3	12.3	13.1
30	27.9	28.1	27.7	27.7	28.2	27.3	28.2
31	22.6 ^a	22.5 ^b	22.5 ^a	22.5 ^a	22.8 ^a	22.2	22.6
32	18.6	25.7	25.6	17.3	18.5	19.0	21.1
AcMe	21.2	21.0	21.0	21.6	21.0	21.0	21.1
AcMe	21.4		21.3	21.7	21.5	21.6	21.3
AcMe	21.5					21.7	21.3
C=O	170.7	170.6	170.6	170.7	170.6	170.6	170.3
C=O	170.9		170.6	170.9	171.2	170.1	170.3
C=O	170.2					170.2	170.5
O–Me						55.2	

*The number of directly attached protons to each individual carbon was verified with the DEPT pulse sequence[25].

† ^{13}C NMR spectrum of **3a** was taken in C_6D_6 .

^{a,b}Assignments may be interchanged in each compound.

(3450 cm^{-1}), ester carbonyl (1740 cm^{-1}) and α - β -unsaturated carboxylic acid (1690 cm^{-1}). In the mass spectrum of **2d**, the molecular ion peak was observed at $m/z\ 570[\text{M}]^+$, suggesting **2d** was a deacetyl ganoderic acid T. This suggestion was consistent with the ^{13}C NMR spectrum of **2d**, which did not contain signals for an acetoxyl methyl and a carbonyl carbon. In the ^{13}C NMR spectrum of **2d** the C-15 carbon signal was shifted upfield to $\delta\ 74.5$ and C-16 carbon signal was shifted downfield to $\delta\ 39.8$. From these spectral investigations, compound **2d** was established to be (22*S*, 24*E*)-3 α ,22-diacetoxy-15 α -hydroxy-5 α -lanosta-7,9(11),24-trien-26-oic acid.

Compound **2e**, ganoderic acid P, mp 211–212.5°, showed in the UV spectrum absorptions at 235, 243, 251 nm, suggesting the presence of a heteroannular diene moiety. The IR spectrum of **2e** showed the presence of hydroxyl (3450 cm^{-1}), ester carbonyl (1735 cm^{-1}) and

α , β -unsaturated carbonyl (1720 cm^{-1}). In the mass spectrum the molecular ion peak was observed at $m/z\ 570[\text{M}]^+$ suggesting that **2e** was a deacetyl ganoderic acid T. In the ^{13}C NMR spectrum of **2e**, the C-3 carbon signal was shifted upfield to $\delta\ 76.1$ and the C-2 carbon signal was shifted downfield to $\delta\ 25.5$. From these data, compound **2e** was established as (22*S*, 24*E*)-15 α ,22-diacetoxy-3 α -hydroxy-5 α -lanosta-7,9(11),24-trien-26-oic acid.

Compound **3b**, *O*-methyl ganoderic acid O, $\text{C}_{37}\text{H}_{56}\text{O}_9$, mp 228–229.5°, showed absorption only at 208 nm (log ϵ 4.1) in the UV spectrum and hydroxyl (3450 cm^{-1}), ester carbonyl (1740 cm^{-1}) and α , β -unsaturated carbonyl (1720 cm^{-1}) absorptions were seen in the IR spectrum. The ^1H NMR spectrum of **3b** (Table 2) clearly showed the presence of one secondary and five tertiary methyl groups, thus suggesting a lanostane skeleton. This was supported by a positive Liebermann–Burchard colour reaction

(brown-yellow). The ^1H NMR and ^{13}C NMR spectral patterns of **3b** were very similar to those of **2a** except for the olefinic region and the appearance of an *O*-methyl signal (δ 55.2). In the ^{13}C NMR spectrum of **3b**, four olefinic carbon signals were observed at δ 129.3 (s, C-25), 132.7 (s, C-9), 139.1 (d, C-24) and 142.9 (s, C-8). On the other hand, six olefinic carbon signals were observed at 115.5 (d, C-11), 121.6 (d, C-7), 129.4 (s, C-25), 139.2 (d, C-24), 140.2 (s, C-8) and 146.2 (s, C-9) in the spectrum of **2a**. The loss of two olefinic carbon signals in the spectrum of **3b** relative to that of **2a** indicated that the diene moiety of **2a** had been replaced by a single double bond. This was supported by the ^1H NMR spectrum, which showed only one olefinic proton signal at δ 6.78 (24-H). These data indicated that **3b** was 7,11-dihydroganoderic acid **T** but with an extra methoxyl group. The position and configuration of the additional methoxyl group of **3b** were elucidated from its ^1H NMR and ^1H - ^1H shift correlation NMR spectral data. In the ^1H NMR spectrum of **3b** a new signal at δ 3.56, which was absent in **2a**, appeared as a broad singlet. The cross peaks between the new signal and the C-6 methylene protons (δ 1.36 and 1.89) were clearly observed.

One of the methylene protons (δ 1.36) coupled with three different protons ($J = 13.5, 13.5, 3.2$ Hz) and another methylene proton only coupled with the geminal methylene proton ($J = 13.5$ Hz). The arrangement from this evidence can only correspond to the $\text{H}_5\text{-H}_{6a}\text{-H}_{6b}\text{-H}_7$ linkages and it permitted the conclusion that the additional methoxyl group was attached at C-7 with an α -configuration. Therefore, the structure of **3b** was elucidated as 7 α -methoxy-7,11-dihydroganoderic acid **T**, (22*S*, 24*E*)-3 α ,15 α ,22-triacetoxy-7 α -methoxy-5 α -lanosta-8,24-dien-26-oic acid.

In the second experiment the benzene extract of the dried cultured mycelia was separated by silica gel column chromatography (see Experimental). After repeating the HPLC, compound **3a**, ganoderic acid **O**, was obtained as colourless needles, mp 156–158°. The molecular formula was established as $\text{C}_{36}\text{H}_{54}\text{O}_9$ by elementary analysis. The UV spectrum of **3a** showed only the presence of an α,β -unsaturated carbonyl group [206 nm (log ϵ 4.1)]. Compound **3a** exhibited hydroxyl (3420 cm^{-1}) and ester carbonyl (1720 cm^{-1}) absorptions in its IR spectrum. The ^1H NMR spectrum of **3a** was very similar to that of **3b** except for the loss of the methoxy methyl proton signal at δ 3.16. ^{13}C NMR spectrum of **3a** was also closely similar to that of **3b** except for the loss of the methoxy methyl carbon signal at δ 55.2 and the up-field shift (δ 67.1) of the methoxy methine carbon signal in the spectrum of **3b**. These data suggested that **3a** had a hydroxyl group instead of a methoxyl group attached to C-7. In order to verify the proposed structure, a mixture of **3a** and methanol was left at room temperature for 7 days. Compound **3b** was detected by HPLC (R_f 16.1 min. Nucleosil C18 85% CH_3CN 3 ml/min.) in the mixture. From these results, the structure of compound **3a**, named ganoderic acid **O**, was established as (22*S*, 24*E*)-3 α ,15 α ,22-triacetoxy-7 α -hydroxy-5 α -lanosta-7,24-dien-26-oic acid. The fact that **3b**, containing a methoxyl group at C-7, was obtained by methanol extraction and was not present in the benzene extract suggests that it is an artifact.

The six new ganoderic acid derivatives identified as major or minor triterpenoid components from the cultured mycelia of *G. lucidum* have 3 α -substituents. In contrast, the ganoderic acid derivatives from the fruit

bodies of *G. lucidum* are 3 β -substituted or 3-keto compounds. In plant tissue cultures of *Isodon japonicus*, a 3 α -hydroxytriterpenoid, 3-epimaslinic acid, is biosynthesized from its 3 β -isomer maslinic acid via a 3-oxo-compound [26]. On the other hand, Rohmer *et al.* [27, 28] reported that in a cell free system from the bacteria *Acetobacter pasteurianum* and *Methylococcus capsulatus*, 3 α - and 3 β -hydroxytriterpenoids were separately formed from (3*R*)- and (3*S*)-2,3-oxidosqualenes, respectively, without involvement of 3-oxo-compounds. The difference in the C-3 substituents between ganoderic acid derivatives isolated from the cultured mycelia and those from the fruiting bodies of *G. lucidum* biosynthetically interesting. We recently observed, using [1,2- $^{13}\text{C}_2$]acetate, that the incorporation pattern of intact bonds of enriched acetate units suggest ganoderic acid **T** to be biosynthesized from (3*S*)-oxidosqualene via a 3-oxo-compound by a process similar to that of in higher plant.

EXPERIMENTAL

Mps were uncorr. Mass spectra were run on a direct insertion probe. NMR spectra were taken in CDCl_3 at 23°; ^{13}C NMR at 75.0 and 100.6 MHz and ^1H NMR at 300 and 400 MHz. HPLC of ganoderic acid derivatives was performed using a Nucleosil C18 (10 \times 300 mm) column, coupled to a UV detector and a differential refractometer.

Isolation and culture conditions of *Ganoderma lucidum*. After sterilizing the cultivated fruiting body of *G. lucidum*, a small piece was inoculated onto a malt agar medium. Developing colonies of the new hyphae from the fruiting body were transferred to malt agar in Petri dishes. The formation of the clamp connection of the pure cultured hyphae was observed under a microscope. The pure mycelium was subcultured for 2 weeks and grown in Roux flasks in 200 ml of medium containing 15 g glucose, 1.0 g soytone, 0.5 g yeast extract, 0.5 g KH_2PO_4 , 0.1 g NaCl, 0.05 g $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$ and 0.11 g $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$ in 1 l of distilled H_2O . The pH of the medium was adjusted to 5.5 with 1 N HCl. Usually, each flask was seeded with 5 of the 10 mm plugs cut from the malt agar culture. The culture was maintained in the dark at 25°.

Extraction procedure and separation of MeOH extract of the mycelia. After 6 weeks culture (100 Roux flasks), the mycelia (5.32 kg, fr. wt) were harvested with Nylon cloth, homogenized with MeOH (10 l) in a Waring blender and allowed to stand for 1 week at room temp. The homogenate was filtered and the residue was re-extracted with the same solvent (10 l). The filtrates were combined and organic solvent was removed under red. pres. The residue was extracted with CHCl_3 (1000 ml \times 2), dried and evapd to dryness. The CHCl_3 extract (16.78 g) was subjected to chromatography over silica gel (800 g Wako gel C-200). Elution with 2 l C_6H_6 and 2 l $\text{C}_6\text{H}_6\text{-EtOAc}$ (9:1) (fraction A), 1 l (fraction B), 4.5 l $\text{C}_6\text{H}_6\text{-EtOAc}$ (8:2) (fraction C), 0.5 l $\text{C}_6\text{H}_6\text{-EtOAc}$ (8:2) and 1 l $\text{C}_6\text{H}_6\text{-EtOAc}$ (6:4) (fraction D) 2 l $\text{C}_6\text{H}_6\text{-EtOAc}$ (6:4) and 2 l EtOAc (fraction E) yielded the crude mixture of compound **1** (fraction B 0.314 g), compounds **2a**, **2b**, **2c** (fraction C 9.332 g) and compounds **2d**, **2e**, **3b** (fraction E 4.35 g).

Isolation and structure elucidation of ergosta-4,7,22-triene-3,6-dione (1). After repeated HPLC (Nucleosil C18, MeOH 3.0 ml/min flow rate, R_f 12.2 min) of fraction B, compound **1** (27.7 mg) was obtained as pale yellow leaflets, mp 156–159° (decomp), $\text{C}_{28}\text{H}_{40}\text{O}_2$ (required 408.3028, $[\text{M}]^+$ at m/z 408.3034); IR $\nu_{\text{max}}^{\text{KBr}}$ cm^{-1} : 2950, 2870 (CH), 1665(CO), 1640(CO), 1620(C=C), 1600 (C=C); UV $\lambda_{\text{max}}^{\text{EtOH}}$ nm (log ϵ): 275 (3.8). ^1H NMR: 0.68 (3H, s), 0.82 (3H, d, $J = 6.6$ Hz), 0.84 (3H, d, $J = 6.6$ Hz), 0.92 (3H, d, $J = 6.6$ Hz), 1.04 (3H, d, $J = 6.6$ Hz), 1.30 (3H, s), 5.14 (1H, dd, $J = 15.0, 6.7$ Hz), 5.24 (1H, dd, $J = 15.0, 7.5$), 5.98 (1H, t, J

= 1.8), 6.47 (1H, s). ^{13}C NMR 75.2 MHz (CDCl_3): 12.8(3), 17.6(3), 19.5(3), 19.7(3), 20.0(3), 21.1(3), 21.9(2), 22.6(2), 27.7(2), 33.1(1), 34.3(2), 35.4(2), 38.5(2), 39.1(0), 40.2(1), 42.9(1), 44.6(0), 47.1(1), 56.1(1), 56.3(1), 123.9(1), 126.0(1), 132.8(1), 134.8(1), 158.2(0), 167.7(0), 187.0(0), 199.3(0). EIMS (direct inlet) 20 eV, m/z (rel. int.): 408 $[\text{M}]^+$ (80), 283 $[\text{M}-\text{side chain}]^+$ (100), 229 $[\text{M}-\text{C}_{13}\text{H}_{23}]^+$ (33), 125 $[\text{M}]^+$ (13).

Isolation and elucidation of ganoderic acid T(2a), ganoderic acid S(2b) and ganoderic acid R(2c). Further separation of fraction C was achieved by HPLC. A small part of fraction C (1.32 g) was run on a Nucleosil C18 packed column (10 mm \times 300 mm) eluted with 95% MeOH (3 ml/min). Compound 2a was isolated from the fraction containing the peak at 16.2 min. Compound 2a (175.4 mg), colourless needles, mp 200–202°, $\text{C}_{36}\text{H}_{52}\text{O}_8$ (required 612.3662, $[\text{M}]^+$ at m/z 612.3666), $[\alpha]_D^{24} + 23^\circ$ (CHCl_3 , c 0.13); IR $\nu_{\text{max}}^{\text{KBr}}$ cm^{-1} : 3420(OH), 2940(CH), 1720(COO), 1700(COO), 1240(C–O); UV $\lambda_{\text{max}}^{\text{EtOH}}$ nm (log ϵ): 214(4.0), 225(4.0), 234(4.0), 242(4.0), 250(3.8). ^1H NMR and ^{13}C NMR: see Tables 2 and 3. EIMS (direct inlet) 20 eV, m/z (rel. int.): 612 $[\text{M}]^+$ (100), 552 $[\text{M}-\text{HOAc}]^+$ (20), 537 $[\text{M}-\text{HOAc}-\text{Me}]^+$ (15), 492 $[\text{M}-2 \times \text{HOAc}]^+$ (18), 477 $[\text{M}-2 \times \text{HOAc}-\text{Me}]^+$ (26), 417 $[\text{M}-3 \times \text{HOAc}-\text{Me}]^+$ (42), 293 $[\text{M}-\text{side chain}-\text{HOAc}-\text{OAc}]^+$ (22), 277 $[\text{M}-\text{side chain}-2 \times \text{HOAc}-\text{Me}]^+$ (20), 251(20), 239(46), 223(18), 197(16), 132(16). Compound 2b was also isolated from the fraction containing the peak at 19.1 min. compound 2b (33.5 mg), colourless needles, mp 194–196°, $\text{C}_{32}\text{H}_{48}\text{O}_8$ (required 512.3502, $[\text{M}]^+$ at m/z 512.3511), $[\alpha]_D^{24} + 19.8^\circ$ (CHCl_3 , c 0.085); IR $\nu_{\text{max}}^{\text{KBr}}$ cm^{-1} : 3410(OH), 2870(CH), 1720(COO), 1675(COO), 1230(C–O); UV $\lambda_{\text{max}}^{\text{EtOH}}$ nm (log ϵ): 214(4.3), 231(4.2), 240(4.2), 248(4.0). ^1H NMR and ^{13}C NMR: see Tables 2 and 3. EIMS (direct inlet) 20 eV, m/z (rel. int.): 512 $[\text{M}]^+$ (100), 479(14), 452 $[\text{M}-\text{HOAc}]^+$ (14), 419(17), 312 $[\text{M}-\text{side chain}]^+$ (30), 311(27), 271(34), 253(31), 171(20), 159(27), 133(16), 119(14), 107(16). Compound 2c was isolated from the last fraction containing the peak at 26.6 min. Compound 2c (22.6 mg), colourless needles, mp 201–202°, $\text{C}_{34}\text{H}_{50}\text{O}_6$ (required 554.3606, $[\text{M}]^+$ at m/z 554.3595), $[\alpha]_D^{24} + 8.7^\circ$ (CHCl_3 , c 0.092); IR $\nu_{\text{max}}^{\text{KBr}}$ cm^{-1} : 3400(OH), 2870(CH), 1720(COO), 1675(COO), 1230(C–O); UV $\lambda_{\text{max}}^{\text{EtOH}}$ nm (log ϵ): 224(4.0), 234(4.0), 242(4.0), 250(3.8). ^1H NMR and ^{13}C NMR: see Tables 2 and 3. EIMS (direct inlet) 20 eV, m/z (rel. int.): 554 $[\text{M}]^+$ (100), 494 $[\text{M}-\text{HOAc}]^+$ (25), 479 $[\text{M}-\text{HOAc}-\text{Me}]^+$ (28), 419 $[\text{M}-2 \times \text{HOAc}-\text{Me}]^+$ (30), 354 $[\text{M}-\text{side chain}]^+$ (24), 313 $[\text{C}_{21}\text{H}_{29}\text{O}_2]^+$ (21), 295(17), 253(53), 239(23), 159(20), 145(16), 133(11), 107(14).

Isolation and elucidation of ganoderic acid Q(2d), ganoderic acid P(2e) and 7-O-methyl ganoderic acid O(3b). Further separation of Fraction E was achieved by repeated HPLC (Nucleosil C18, solvent 85% CH_3CN , flow rate 3.0 ml/min.) and compounds 2d, 2e and 3b were isolated from the fraction b (21.3 mg) containing the peak at 15.9 min, the fraction a (8.9 mg) containing the peak at 15.0 min and the fraction c (27.9 mg) containing the peak at 17.6 min respectively. Compound 2d, mp 131–132°, $\text{C}_{34}\text{H}_{50}\text{O}_7$ (required 570.3555, $[\text{M}]^+$ at m/z 570.3568); IR $\nu_{\text{max}}^{\text{KBr}}$ cm^{-1} : 3450(OH), 2950(CH), 1740(COO), 1690(CO), 1250; UV $\lambda_{\text{max}}^{\text{EtOH}}$ nm (log ϵ): 216(4.3), 225(4.2), 235(4.3), 243(4.3), 251(4.1). ^1H NMR and ^{13}C NMR: see Tables 2 and 3. EIMS (direct inlet) 20 eV, m/z (rel. int.): 570 $[\text{M}]^+$ (100), 510 $[\text{M}-\text{HOAc}]^+$ (35), 495 $[\text{M}-\text{HOAc}-\text{Me}]^+$ (44), 435 $[\text{M}-\text{HOAc} \times \text{Me}]^+$ (19), 417 $[\text{M}-\text{HOAc} \times 2-\text{H}_2\text{O}-\text{Me}]^+$ (25). Compound 2e, mp 211–212.5°, $\text{C}_{34}\text{H}_{50}\text{O}_7$ (required 570.3555, $[\text{M}]^+$ at m/z 570.3552), IR $\nu_{\text{max}}^{\text{KBr}}$ cm^{-1} : 3450(OH), 2950(CH), 1735(COO), 1720(COO), 1250(C–O), UV $\lambda_{\text{max}}^{\text{EtOH}}$ nm (log ϵ): 214(4.3), 225(4.2), 235(4.2), 243(4.3), 251(4.1). ^1H NMR and ^{13}C NMR: see Tables 2 and 3. EIMS (direct inlet) 20 eV, m/z (rel. int.): 570 $[\text{M}]^+$ (100), 510 $[\text{M}-\text{HOAc}]^+$ (23), 495 $[\text{M}-\text{HOAc}-\text{Me}]^+$ (39), 435 $[\text{M}-2$

$\times \text{HOAc}-\text{Me}]^+$ (12), 417 $[\text{M}-2 \times \text{HOAc}-\text{H}_2\text{O}-\text{Me}]^+$ (12). Compound 3b, mp 228–229.5°, $\text{C}_{37}\text{H}_{56}\text{O}_9$ (required 644.3924, $[\text{M}]^+$ at m/z 644.3895); IR $\nu_{\text{max}}^{\text{KBr}}$ cm^{-1} : 3450(OH), 2970, 2950(CH), 1730(COO), 1720(COO), 1250(C–O); UV $\lambda_{\text{max}}^{\text{EtOH}}$ nm (log ϵ): 208(4.1). ^1H NMR and ^{13}C NMR: see Tables 2 and 3. EIMS (direct inlet) 20 eV, m/z (rel. int.): 644 $[\text{M}]^+$ (3), 612 $[\text{M}-\text{HOME}]^+$ (17), 584 $[\text{M}-\text{HOAc}]^+$ (79), 537 $[\text{M}-\text{HOAc}-\text{Me}]^+$ (17), 419 $[\text{M}-3 \times \text{HOAc}-\text{HCHO}-\text{Me}]^+$ (100).

Synthesis of compound 2f. After methylation of compound 2a (9.8 mg) with CH_3N_2 , methylated compound 2a was reacted with LiAlH_4 in 1.5 ml THF for 5 min at room temp. The reaction was stopped with 5 drops of 10% NH_4Cl . After removing of THF, the residue was partitioned with H_2O and CHCl_3 . The CHCl_3 fraction was recrystallized from $\text{H}_2\text{O}-\text{MeOH}$ to give colourless plates (1.78 mg), mp 201–202°, $\text{C}_{30}\text{H}_{48}\text{O}_4$ (required 472.3552, $[\text{M}]^+$ at m/z 472.3545); CD curve $[\theta]_{239}^{25} + 12689$ EtOH; $c = 0.012$. ^1H NMR (400 MHz, CDCl_3): δ 0.63 (3H, s), 0.89 (3H, d, $J = 6.5$ Hz), 0.98 (6H, s), 1.00 (3H, s), 1.40 (1H, m), 1.51 (1H, dd, $J = 10.5, 5.0$ Hz), 1.69 (3H, s), 3.45 (1H, br s), 3.66 (1H, br dd, $J = 9.0, 5.0$ Hz), 4.04 (2H, s), 4.28 (1H, br t, $J = 7$ Hz), 5.34 (1H, br d, $J = 6.2$ Hz), 5.46 (1H, br t, $J = 7$ Hz), 5.85 (1H, br d, $J = 6$ Hz). ^{13}C NMR (100.6 MHz, CDCl_3): δ 11.8(3), 14.0(3), 15.9(3), 17.4(3), 22.7(3), 22.8(3), 22.9(2), 25.6(2), 28.2(3), 29.9(2), 33.9(2), 37.3(0), 37.4(0), 38.6(2), 39.5(2), 40.5(1), 43.1(1), 44.3(0), 45.3(1), 52.1(0), 68.7(2), 73.3(1), 74.8(1), 76.1(1), 115.7(1), 121.4(1), 122.1(1), 137.6(0), 140.8(0), 146.2(0).

Extraction procedure and separation of the C_6H_6 extract. After 42 days culture (97 Roux flasks), the mycelia (fr. wt 5.64 kg) were lyophilized and the dried mycelia (147 g) were extracted with $\times 2$ C_6H_6 (total 12 l) at room temp. After filtration, the C_6H_6 extract (10.5 g) was obtained. A part (6.1 g) of the benzene extract was subjected to chromatography over silica gel (500 g Wako gel C-200) and eluted as follows: fraction 1, 1 l $\text{C}_6\text{H}_6-\text{Me}_2\text{CO}$ (9:1) and 2.1 l $\text{C}_6\text{H}_6-\text{Me}_2\text{CO}$ (1:3); fraction 2, 1.9 l $\text{C}_6\text{H}_6-\text{Me}_2\text{CO}$ 0.2 l $\text{C}_6\text{H}_6-\text{Me}_2\text{CO}$ (8:2); fraction 3, 0.6 l $\text{C}_6\text{H}_6-\text{Me}_2\text{CO}$ (8:2); fraction 4, 1.55 l $\text{C}_6\text{H}_6-\text{Me}_2\text{CO}$ (8:2); fraction 5, 3.05 l $\text{C}_6\text{H}_6-\text{Me}_2\text{CO}$ (8:2) and 1 l $\text{C}_6\text{H}_6-\text{Me}_2\text{CO}$ (50:50). Fractions 1 and 2 contained the mixture of compounds 2a, 2b, 2c (2.07 g) and compound 3a (1.65 g), respectively.

Isolation and structure elucidation of compound 3a. Further purification of fraction 2 was achieved by HPLC. A part (951.5 mg) of fraction 2 was run on Nucleosil C18 packed column (10 \times 300 mm) eluted with $\text{CH}_3\text{CN}-\text{H}_2\text{O}$ 17:33 ml/min. Compound 3a was obtained from the fraction (214.0 mg) containing the peak at 12.7 min. but it was still impure and further purification was carried out with different HPLC condition (Nucleosil C18, solvent $\text{CH}_3\text{CN}-\text{H}_2\text{O}$ 3:1 3.0 ml/min.) The pure compound 3a (128.6 mg) colourless needles, had mp 156–158° (Found: C, 68.45; H, 8.68, $\text{C}_{36}\text{H}_{54}\text{O}_9$ requires: C, 68.54; H, 8.63). IR $\nu_{\text{max}}^{\text{KBr}}$ cm^{-1} : 3450(OH), 2980(CH), 1735(COO), 1260(C–O); UV $\lambda_{\text{max}}^{\text{EtOH}}$ nm (log ϵ): 206(4.1). ^1H NMR and ^{13}C NMR: see Tables 2 and 3. EIMS (direct inlet) 20 eV, m/z (rel. int.): 612 $[\text{M}-\text{H}_2\text{O}]^+$ (17), 570 $[\text{M}-\text{HOAc}]^+$ (36), 537 $[\text{M}-\text{HOAc}-\text{H}_2\text{O}-\text{Me}]^+$ (31), 495 $[\text{M}-\text{HOAc} \times 2-\text{Me}]^+$ (100), 435 $[\text{M}-\text{HOAc} \times 3-\text{Me}]^+$ (10), 417 $[\text{M}-\text{HOAc} \times 3-\text{Me}-\text{H}_2\text{O}]^+$ (12), 239 (18), 171(9).

X-ray data of compound 2a. Atomic coordinates have been deposited at the Cambridge Crystallographic Data Centre.

The conditions for the NMR measurements. The homo and hetero nuclear two-dimensional chemical shift correlation diagram was obtained according to the method described in a previous paper [4].

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